Defects in Cu(In,Ga)Se₂ Chalcopyrite Semiconductors: A Comparative Study of Material Properties, Defect States, and Photovoltaic Performance

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Understanding defects in Cu(In,Ga)(Se,S)₂ (CIGS), especially correlating changes in the film formation process with differences in material properties, photovoltaic (PV) device performance, and defect levels extracted from admittance spectroscopy, is a critical but challenging undertaking due to the complex nature of this polycrystalline compound semiconductor. Here we present a systematic comparative study wherein varying defect density levels in CIGS films were intentionally induced by growing CIGS grains using different selenium activity levels. Material characterization results by techniques including X-ray diffraction, scanning electron microscopy, transmission electron microscopy, secondary ion mass spectrometry, X-ray photoelectron spectroscopy, and medium energy ion scattering indicate that this process variation, although not significantly affecting CIGS grain structure, crystal orientation, or bulk composition, leads to enhanced formation of a defective chalcopyrite layer with high density of indium or gallium at copper antisite defects ([(In, Ga)₉₀] near the CIGS surface, for CIGS films grown with insufficient selenium supply. This defective layer or the film growth conditions associated with it is further linked with observed current-voltage characteristics, including rollover and crossover behavior, and a defect state at around 110 meV (generally denoted as the N₁ defect) commonly observed in admittance spectroscopy. The impact of the [(In, Ga)₉₀] defects on device PV performance is also established.

1. Introduction

Photovoltaic (PV) technologies could play a vital role in satisfying human being’s ever-increasing demand for energy through directly converting solar radiation, one of the cleanest and most abundant energy sources on earth, to electricity.¹² Cu(In,Ga)(Se,S)₂ (CIGS), with attributes including 1) high optical absorption as a direct band-gap semiconductor,⁵ 2) electrically benign grain boundaries due to large atomic relaxation, and ³ reduced usage of the toxic element cadmium (i.e., compared with CdTe solar cells),⁶ is an attractive material candidate for the absorber layer in PV devices. Record laboratory-scale power conversion efficiency is above 20% for CIGS PV devices,⁷ the highest among all thin-film type solar cells. Although CIGS films in record devices are deposited by a vacuum-based co-evaporation process, non-vacuum-based deposition techniques, such as chalcogenide-hydrazine complex precursors,⁷–⁹ oxide/chalcogenide nanoparticle precursors,¹⁰–¹³ and electrodeposition,¹⁴ are attracting increasing attention due to their potential to achieve lower cost and larger volume manufacturing, with advantages including 1) lower equipment capital costs, 2) higher throughput, 3) more convenient scale-up potential for large area processing, and 4) higher materials utilization rate, compared with vacuum based deposition techniques.¹⁶,¹⁷

Despite the great progress made on optimizing the deposition process and PV performance of CIGS devices, the development of this material and associated PV technology has been accomplished mainly on an empirical basis, where improvements have been found by trial and error. Knowledge about defects in CIGS, especially their correlation with process conditions and material properties as well as their influences on device performance, is still incomplete.¹⁸ Controversies have surrounded the interpretation of defect levels observed in admittance spectroscopy and phenomena such as the partial saturation of the device forward current (rollover) and crossing of dark and illuminated current density-voltage (J–V) curves (crossover) observed in current-voltage (I–V) characteristics of CIGS solar cells, whose origins have been attributed to Fermi-level pining at the interface,²⁰ bulk defects in CIGS,²²,²³ non-Ohmic contact between CIGS and Mo,²⁴–²⁶ deep acceptor defects in the CdS buffer layer,²⁷–³⁰ and the existence of a defective chalcopyrite layer, sometimes referred to as an order defect compound (ODC) phase, close to the heterojunction.³¹–³⁸

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Here we report a systematic comparative study of CIGS material and PV devices, aiming at correlating the electrical defect levels of CIGS observed in admittance spectroscopy with structural defects as measured by X-ray diffraction (XRD), scanning electron microscopy (SEM), transmission electron microscopy (TEM), secondary ion mass spectrometry (SIMS), X-ray photoelectron spectroscopy (XPS), and medium energy ion scattering (MEIS), and clarifying their influences on device performance characterized via optoelectrical and quantum efficiency (QE) measurements. Based on these experimental results, the characteristic admittance step at around 110 meV, which is often referred to as the N1 defect, is associated with a defective chalcopyrite layer identified with material characterization techniques, which has high density of (In, Ga) on Cu antisite defects ((In, Ga)Cu) and is close to the heterointerface. This defective layer is further connected with unusual I–V signatures like rollover and crossover commonly observed in CIGS solar cells. These (In, Ga)Cu defects, which can be converted from shallow donors into deep level electron traps under photoexcitation, serve as active recombination centers, as evident from QE measurement, and therefore lead to inferior PV performance of associated solar cells. The above model allows a unified explanation of material properties, some unusual I–V characteristics, defect levels extracted from admittance spectroscopy and PV performance observed in CIGS solar cells, and is expected to be important for future development/optimization of PV-targeted CIGS deposition processes.

2. Results and Discussion

2.1. CIGS Film Growth Conditions

CIGS films were formed by a liquid-based spin-casting process, using a hydrazine solution of metal chalcogenide precursors as a representative method, to study the defects in CIGS films induced by different process conditions. Experimental details of this process have been described in previous publications. Briefly, CIGS precursor films with a fixed CuIn0.92Ga0.08Se2 stoichiometry and identical initial Se/metal ratio of ca. ~1.3 were deposited on Mo-coated soda-lime glass substrates. These films were then annealed with rapid thermal annealing (RTA) at a final temperature of 600 °C in nitrogen atmosphere to transform as deposited nano-crystalline precursors into polycrystalline CIGS. The selenium activity in the films during the final high-temperature heat treatment was varied by changing the ramp profile during the RTA process (four heating profiles tested, denoted as Max., High, Low, Min., respectively, with descending temperature ramp rate). Lower ramping-up speed reduces the Se supply level during the grain growth process, as it increases Se loss in the temperature range where the temperature is sufficiently high to evaporate Se from the precursor film but insufficient to induce CIGS grain growth. Lower Se supply levels during the CIGS grain growth process may lead to the formation of more structural defects in the CIGS lattice due to Se deficiency, especially considering the high vapor pressure, ca. ~0.26 atm at 600 °C[41] of Se under the processing conditions employed. Other components of completed CIGS solar cells, such as Mo, CdS, ZnO, and tin-doped indium oxide (ITO) layers, were all deposited under the same conditions, which excludes their possible impacts on PV performance and defect spectra of those devices, and therefore makes the quality of CIGS films as the only experimental variable accounting for differences observed in the following comparative studies.

2.2. PV Device Characterizations

Figure 1a summarizes the comparison of power conversion efficiencies of solar cells based on CIGS films grown with different Se supply levels. A statistically significant increase of device efficiency with the increase of Se supply is observed. Figure 1b shows the illuminated J–V characteristics of associated typical CIGS devices. Devices based on CIGS films grown with higher Se supply demonstrate improved device characteristics in all aspects, including open circuit voltage (VOC), short circuit current (JSC), and fill factor (FF). The measurement of internal quantum efficiencies (IQE) (Figure 1c) further reveals
that the differences in device performance are mainly caused by stronger recombination in the CIGS layer grown with lower Se supply levels, as evident from reduction of their IQE in the long wavelength regime (λ > 550 nm). The band gap energies (E_g) of these CIGS films are also determined from their IQE spectra (Figure 1d). Although these CIGS films demonstrate drastically different PV characteristics and different levels of carrier recombination, they have almost the same E_g. These results suggest that the lower performance of devices based on CIGS films grown with low Se supply levels are most likely caused by the increased formation of defects in CIGS, which act as active recombination centers, instead of differences in CIGS band gaps or bulk compositions.

2.3. CIGS Film Characterizations

To clarify the material origin of these electrically active defects in CIGS, the crystal orientation, microstructure, bulk composition, and surface composition of CIGS films grown with different Se supply levels were carefully characterized. The crystal orientations of CIGS and MoSe_2 formed at the CIGS-Mo interface were recorded by XRD as depicted in Figure 2a. It has been demonstrated in vacuum-based deposition approaches that very high, ca. >10, Se/metal flux ratios can promote the growth of CIGS along the <220> orientation, as evident from ca. >5 times decrease of the (112)/(220) peak ratio, which is believed to produce a lower density of non-radiative recombination centers compared with the more typical <112> orientation. Here we are looking at a more subtle difference in Se supply levels. In our Se/metal supply ratio range, the CIGS (112)/(220) peak ratios fluctuate at around 2.5, indicating random crystal orientation of CIGS grains and that this fluctuation cannot be meaningfully correlated with the modulation of Se supply level during CIGS grain growth (Figure 2b). So we conclude that the crystal orientation of the bulk CIGS film is unaffected by the change of process conditions in our experiments and therefore that this parameter does not account for the differences in PV performance and defect state densities.

The crystal orientation of the MoSe_2 layer formed at the CIGS-Mo interface is also important. It has been reported that the formation of a MoSe_2 layer with correct crystal orientation improves the adhesion of CIGS on Mo and is critical for the formation of ohmic contacts between CIGS and the Mo electrode. Under all process conditions examined in this study, only peaks associated with the <100> and <110> orientations of MoSe_2 are observed, which implies that the MoSe_2 layers are grown with c axis parallel with the Mo substrate. This perpendicularly oriented layer is desirable for better adhesion and electrical contact between CIGS and Mo.

The peak intensities associated with MoSe_2 are significantly enhanced in XRD patterns of CIGS films grown with higher Se supply, which suggests the increase of MoSe_2 layer thickness with the increase of Se supply levels during CIGS grain growth, as has been previously reported.

To confirm this increase of MoSe_2 layer thickness with the increase of Se supply level and, more importantly, to examine the grain structures of CIGS films, cross-section micrographs of CIGS films grown with Min. and Max. Se supply levels were obtained with both SEM and TEM as shown in Figure 3. CIGS film microstructures, especially CIGS grain size, are important for device performance, as smaller grains may lead to both unfavorable hopping based electron transport and stronger recombination at grain boundaries. However, SEM images (Figure 3a and 3c) show that the small difference in Se supply levels, as designed in our experiments, has little impact on CIGS grain structures. The small reduction in grain size of CIGS films grown with lower Se supply levels is insufficient to explain their dramatically different, i.e. 5.1 ± 0.3% (Min.) versus 10.8 ± 0.7% (Max.), power conversion efficiencies (Figure 1a). The crystal orientation of the MoSe_2 layer at the CIGS-Mo interface under higher Se supply is directly observed, and becomes more evident in the TEM cross sections (i.e., compare Figure 3b and 3d).

To test whether the difference in MoSe_2 thickness in this range, i.e., 50 – 300 nm, will affect the PV performance of CIGS...
solar cells or not, we did a control experiment where the thickness of the MoSe\textsubscript{2} layer for CIGS films grown with Max. Se supply was significantly reduced to a level close to that of CIGS films grown with Min. Se supply, i.e. from \(\sim 300\) nm to \(\sim 80\) nm. This reduction in MoSe\textsubscript{2} thickness was accomplished via sputtering a thin (i.e., 60 nm) layer of zero stress high-density Mo on top of the typical lower-density Mo, to block the diffusion of Se into Mo during the high temperature CIGS grain growth process (Figure 3b and 3c).\footnote{51} No statistically significant difference in device PV performance is observed with this reduction of MoSe\textsubscript{2} layer thickness (Figure 3a). Based on these results, we draw the conclusion that the observed differences in PV performance should not be correlated with either CIGS microstructures or the CIGS-Mo back contact, which is mediated by MoSe\textsubscript{2} layers with identical favorable crystal orientation and within certain thickness limits.

The composition profiles of these CIGS films were characterized with SIMS (Figure 5). As expected, the films have identical bulk compositions and this conclusion is also supported by the IQE measurements, which indicate that each film has the same \(E_G\) (Figure 1d). The formation of a thick MoSe\textsubscript{2} layer at the CIGS-Mo interface for CIGS films grown with high Se supply is also evident from the SIMS depth profile of Se, where a broad peak of high concentration of Se near the back contact is observed. Although the SIMS technique is very useful for analyzing CIGS composition, with advantages including extreme sensitivity, high mass resolution, and wide dynamic range, the depth resolution is limited by the film surface roughness, which is above 15 nm (root mean square) as measured by atomic force microscopy (AFM). Therefore, additional analytical techniques are necessary to study the surface composition of CIGS, which is very critical for PV performance due to proximity to the heterojunction.\footnote{52}

The surface compositions of pristine CIGS films grown with Min. and Max. Se supply levels were compared based on XPS measurement results. Contrary to previous results obtained with a three-stage co-evaporation process,\footnote{53,54} no Cu\textsubscript{2}Se phase segregation near the surface is identified for CIGS films grown with low Se supply, as evident from no significant difference for Cu\textsubscript{2}Se, In\textsubscript{2}Se\textsubscript{3}, and Se\textsubscript{4} peak intensities, possibly due to different reaction chemistry of our non-vacuum based approach, which has pre-formed metal-chalcogenide bonds in the precursors. However, almost two times higher Ga content near the top
surface for CIGS films grown with Min. Se supply after comparing both the Ga LMM Auger and Ga\(^{3p}\) peaks with those of CIGS films grown with Max. Se supply (Figure 6). To get more quantitative information about surface composition of CIGS films, MEIS analysis was performed. MEIS is capable of providing quantitative information about elemental composition of materials with 5 – 10 nm probe depth by measuring the energy spectrum of backscattered He\(^{+}\) ions.\[55\] For CIGS films grown with high Se supply, the surface composition is almost identical to the bulk composition as measured from quantitative SIMS and deduced from ratios among components in the precursor solution (target composition). However, for CIGS films grown with low Se supply, the Se atomic percent near the top surface decreases significantly (Table 1). This reduction of anions from the CIGS lattice makes the top surface become cation-rich, especially (In, Ga) rich considering that the bulk composition of our CIGS films is Cu poor. In order to accommodate these extra (In, Ga) atoms, one possibility is to form more Se vacancies \(V_{Se}\). In that case, (In, Ga) to Cu ratio should remain the same as in the bulk. The other possible route is to let these extra (In, Ga) atoms occupy Cu vacancies \(V_{Cu}\) to form \(\text{(In, Ga)Cu}^0\) where an increase of the (In, Ga) to Cu ratio in the top layer is expected due to the diffusion of additional \(V_{Cu}\) to the top surface from the bulk. \textit{Ab initio} simulation predicts lower formation energy of \(\text{(In, Ga)Cu}^0\) compared with that of \(V_{Se}\), and therefore the second route is more favorable in terms of energy.\[56\] This prediction is supported by our experimental results, where the (In, Ga) to Cu ratio increases, mainly due to the increase of Ga content, for the surface layer of CIGS films grown with low Se supply. Similar behaviors, although to a lesser extent, are also present in the CIGS films grown with high Se supply after subjecting to a 5 min high temperature (ca. 650 °C) annealing in the MEIS system vacuum chamber to evaporate some amount of Se atoms from the CIGS lattice (Table 1). In addition, both XPS and MEIS results point out that Ga atoms are more likely to occupy \(V_{Cu}\) perhaps since their atomic radius (ca. 1.30 Å) is closer to that of Cu (ca. 1.35 Å) compared with \((\text{In, Ga})\) \(V_{Cu}\) (ca. 1.55 Å).\[57\] The formation of this defective chalcopyrite layer with high density of \(\text{(In, Ga)Cu}^0\) defects has significant impact on device PV performance and defect spectra, as will be discussed in Section 2.4.

### 2.4. Current-Voltage Signatures and Defect Spectra from Admittance Measurement

Simulation results suggest that these \(\text{(In, Ga)Cu}^0\) defects, which behave as shallow donors in the ground state, can capture two electrons to assume an off-centre atomic configuration and become deep defect-localized states with a substantial energy barrier, ca. –0.4 eV, for trapped electrons to re-emit into the conduction band, under photoexcitation\[58\]

\[
\text{(In, Ga)Cu}^2^+ + 2e \leftrightarrow \text{(In, Ga)Cu}^0. \tag{1}
\]

Then, they can rapidly capture free holes to back transit into the substitutional configuration,

\[
\text{(In, Ga)Cu}^0 + h \leftrightarrow \text{(In, Ga)Cu}^{2+} + e. \tag{2}
\]

creating a channel for non-radiative recombination, following the standard Shockley-Read-Hall (SRH) recombination mechanism.\[18\] On the other hand, since this defective

### Table 1. Comparison of bulk elemental compositions of CIGS, as measured by the mole ratio of precursors (Target) and quantitative SIMS, with surface compositions, as measured by medium energy ion scattering (MEIS), of CIGS films grown with Min. and Max. Se supply levels, as well as the change of surface composition of CIGS film grown with Max. Se supply after annealing at 650 °C for 5 min in vacuum.

<table>
<thead>
<tr>
<th></th>
<th>Cu</th>
<th>(In,Ga)</th>
<th>Se</th>
<th>(In,Ga)/Cu</th>
</tr>
</thead>
<tbody>
<tr>
<td>Target</td>
<td>23.5</td>
<td>25.5</td>
<td>51.0</td>
<td>1.08</td>
</tr>
<tr>
<td>SIMS</td>
<td>23.9</td>
<td>26.9</td>
<td>47.5</td>
<td>1.12</td>
</tr>
<tr>
<td>MEIS (Max. Se)</td>
<td>25.0</td>
<td>26.4</td>
<td>46.0</td>
<td>1.06</td>
</tr>
<tr>
<td>MEIS (Max. Se + Anneal)</td>
<td>25.0</td>
<td>29.0</td>
<td>44.0</td>
<td>1.16</td>
</tr>
<tr>
<td>MEIS (Min. Se)</td>
<td>23.3</td>
<td>28.1</td>
<td>41.4</td>
<td>1.21</td>
</tr>
</tbody>
</table>

Figure 5. Comparison of secondary-ion mass spectroscopy (SIMS) depth profiles for major elements of CIGS including Cu (a), In (b), Ga (c), and Se (d) for completed devices based on CIGS films grown with Min. Se supply (black dash-dotted line) and Max. Se supply (red solid line), respectively.

Figure 6. Comparison of (a) Ga LMM Auger peak intensities and (b) Ga\(^{3p}\) peak intensities for CIGS films grown with Min. (dash dotted line) and Max. (solid line) Se supply levels.
The defective chalcopyrite layer near the heterojunction, among V Se donors in the CIGS absorber, or a structural defects formed at the CIGS top surface acts as a potential barrier at the back contact, defect states in the front CdS window layer, or a defective chalcopyrite layer close to the heterojunction, standard SRH recombination, which is frozen out at T = 0 K, dominates in our devices. In addition to the high temperature linear regime, there is a low temperature saturation regime where \( V_{OC} \) may become approximately temperature independent. This saturated \( V_{OC} \), which is obtained after freezing out the major recombination mechanism, indicates the maximum splitting of the quasi-Fermi levels, i.e. built-in potential of the p-n junction, and is correlated with doping concentration in CIGS or Fermi level pinning. \( V_{OC} \) of the device based on a CIGS film grown with Min. Se level saturates at higher temperature (~200 K), agreeing well with the reduction of doping level caused by the occupation of the very shallow acceptor V Cu by (In, Ga) observed in XPS and MEIS.

After clarifying the material origin and recombination mechanism of these defects, they are further related with unusual J–V characteristics and defect states commonly observed in CIGS solar cells. Illuminated and dark J–V curves of devices based on CIGS films grown with Min. and Max. Se supply levels measured at temperature between 100 and 300 K are shown in Figure 7b and 7c. For devices based on CIGS films grown with low Se supply, two characteristic features, including the partial saturation of the device forward current (rollover) and crossing of dark and illuminated J–V curves (crossover), are obvious and become more severe at lower temperatures. However, these signatures are much less pronounced for devices based on CIGS films grown with high Se supply. Both rollover and crossover behaviors are commonly observed in many CIGS solar cells and their origin has been correlated with a Schottky barrier at the back contact.\(^{24–26}\) Defect states in the front CdS window layer,\(^{27–30}\) V Se donors in the CIGS absorber,\(^{61–63}\) or a defective chalcopyrite layer near the heterojunction,\(^{34–36}\) among which the explanation based on defective layer formation is supported by our material characterization results. According to device simulation, this thin layer with high density of structural defects formed at the CIGS top surface acts as a potential barrier and causes a significant potential drop across this layer, which leads to the rollover behavior.\(^{25}\) The presence of this potential barrier also increases the series resistance and reduces the fill factor of the device. These defects can be neutralized by photogenerated carriers under illumination, which enables reduced potential drop in this defective top layer.\(^{34}\) Therefore, higher current output for illuminated J–V curves than that of dark J–V, or the crossover behavior, is observed.

Another controversial issue in CIGS is the interpretation of the defect density spectra obtained from admittance spectroscopy, especially their correlation with defect structures and PV performance. Figure 8a shows the defect level distribution for the CIGS film grown with high Se supply level. Two trap

\[ V_{OC} = \frac{E_A}{q} - \frac{n k T}{q} \ln \left( \frac{J_{0A}}{J_f} \right). \]
3. Conclusions
We summarize our results as follows: the Se supply level during CIGS grain growth is controlled to induce different levels of structural defects in CIGS films, while keeping all other process parameters identical to simplify the data interpretation. This process difference has negligible effect on CIGS grain structures, crystal orientation, and bulk composition, as measured by cross-section SEM/TEM, XRD and SIMS. However, under insufficient Se supply level, a defective chalcopyrite layer with high density of (In, Ga)\textsubscript{Cu} defects forms near the CIGS surface, as suggested by XPS and MEIS results, due to the loss of Se from the CIGS lattice. This defective chalcopyrite layer leads to more pronounced rollover and crossover signatures observed in $J$–$V$ curves. The occupation of existing V\textsubscript{Cu} defects by extra (In, Ga) atoms is also reflected in their defect spectra, where significant decrease of shallow acceptor defects, attributed to V\textsubscript{Cu} defects, but increase of the deeper N1 defect level, attributed to (In, Ga)\textsubscript{Cu} defects, are observed, compared with results from the same measurements performed on CIGS films grown with high Se supply. The direct and significant negative impact of those defects on solar cell PV performance by acting as SRH recombination centers is also established. These systematic comparative study results are not only important for understanding the materials science and defect physics of CIGS, especially clarifying the controversies about the origin of the unusual shape of the $J$–$V$ characteristic and N1 defect level in admittance spectroscopy, but also useful for the further optimization of CIGS process conditions for better PV device performance.

4. Experimental Section

**PV Device Fabrication:** CIGS films were deposited from solution on Mo-coated glass substrates as previously described, followed by a CdS layer (~60 nm thick), deposited with a standard chemical bath approach at 60 °C to form the heterojunction. Intrinsic ZnO (80 nm) and ITO (140 nm) were sputtered on top with a commercial RF magnetron sputtering system (Kurt Lesker). Finally, a Ni (50 nm)/Al (2 μm) grid was patterned with shadowed e-beam evaporation to facilitate carrier collection. No anti-reflective coating was employed. Each device was separated using mechanical scribing, defining a device with an approximate area of 0.45 cm$^2$.

**Film Characterization:** The grain structures of CIGS films were characterized with cross-section SEM (Zeiss, Leo-1560) operated at 5 kV, after mechanical cleavage and a thin Au film coating to prevent charging. TEM was performed using a JEM 3000F operated at 300 kV. Cross-sectioned samples were prepared using the FEI Helios 400S DB-FIB. XRD data were collected with a Siemens D5000 diffractometer using Cu Ka radiation. The comparative SIMS measurements were performed with a Cameca SC-Ultra instrument with cesium ion bombardment and quantitative SIMS measurements were performed at Evans Analytical Group as a service. CIGS surface roughness was measured with an AFM (Dimension 3000) in tapping mode. Sample surface compositions were measured in the near-surface region using XPS and MEIS. The XPS system used a Mg Ka source, with photoelectrons collected at a 45 degree emission angle, giving an escape depth of approximately 1.5 nm. MEIS measurements were performed with 200 keV He$^+$ ions, using a toroidal electrostatic analyzer for detection of backscattered ions.

**Device Characterization:** The completed PV devices were characterized in air with a source meter (Keithley 2440) and a 1000 W Xe lamp solar simulator with 1.5 air mass filter (Oriel 91192) equipped with closed-loop light stabilization system. The light source was calibrated to 100 mW cm$^{-2}$ with a reference cell (Oriel 91150) each time before measurement. The capacitance was measured with an LCR meter (Hewlett-Packard 4192A). Temperature dependent measurement was carried out using a liquid...
nitrogen (LN2) cooled cryostat and a temperature controller (Lake Shore Cryotronics DRC-91C). QE and Reflectivity were measured with Xe light source, monochromator, light chopper and lock-in amplifier system (Protolux QE 1400).

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